

Asymmetric Synthesis of the Diastereoisomers of L-1-Indanylglycine and L-1-Benz[f]indanylglycine, χ_1, χ_2 -Constrained Side-Chain Derivatives of L-Phenylalanine and L-2-Naphthylalanine

Hubert Josien and Gérard Chassaing*

Laboratoire de Chimie Organique Biologique, U.A. C.N.R.S. 493, Université Pierre-et-Marie Curie, Paris, France

(Received 3 August 1992; accepted 21 September 1992)

Key Words: side-chain constrained α amino acids; L-1-indanylglycines; L-1-benz[f]indanylglycines.

Abstract: The diastereoisomers of L-1-indanylglycine and L-1-benz[f]indanylglycine, novel topographic tools and analogues of phenylalanine and 2-naphthylalanine, were synthesized from a sultam-derived glycine imine synthon alkylated by judicious electrophiles, and subsequent hydrolysis and sultam-cleavage. An X-ray analysis on one alkylation product established the β -configuration.

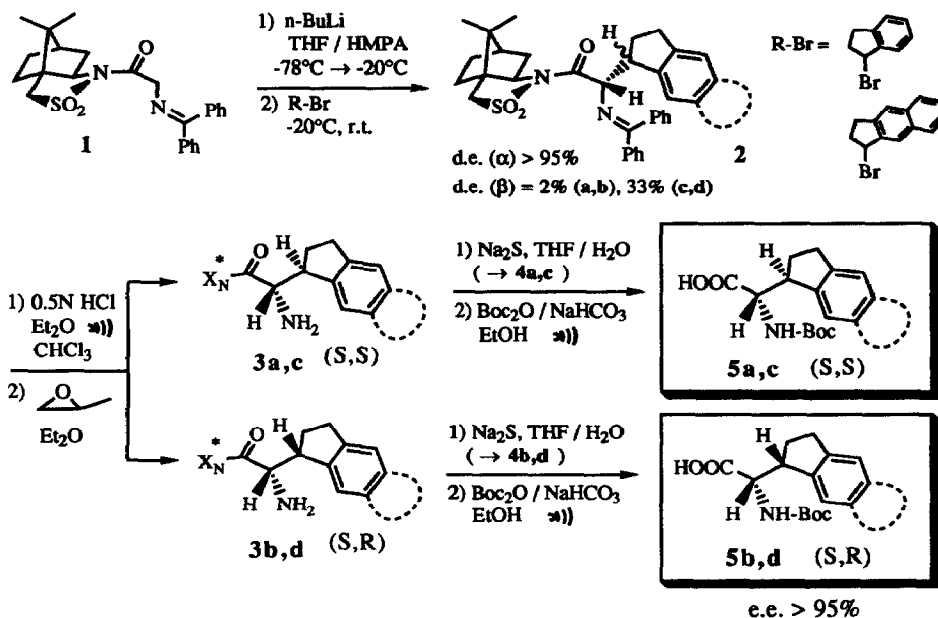
The determination of the bioactive conformation of a peptide is an important step toward the design of new pharmaceutical drugs. This analysis is usually hampered by the flexibility of the backbone and the side-chains. Thus, a general approach involves the use of conformational constraints to stabilize firstly the backbone conformation, secondly to control the side-chain orientation. The side-chain can adopt three major conformers from the α -carbon: gauche (+), gauche (-) or trans ($\chi_1 = +60^\circ, -60^\circ$, or 180°). We previously described the synthesis of two bis-aromatic probes, L-diphenylalanine, and L-9-fluorenylglycine^{5b} which allow the simultaneous stabilization of two phenylalanine rotamers. However, to determine if one of these rotamers is favored in the receptor site, the introduction of a second chiral center on the β -carbon is necessary. This purpose was achieved by Hruby *et al.* with the diastereoisomers of β -methyl-phenylalanine¹. We propose the synthesis of new topographic probes: the diastereoisomers of L-1-indanylglycine (Ing) and L-1-benz[f]indanylglycine (Bfi) **4a-d**, constrained side-chain derivatives of phenylalanine, and 2-naphthylalanine, respectively. These amino acids are more constrained than the β -methyl-phenylalanines: χ_1 is still controlled by the chiral center on the β -carbon, but the cyclopentene ring imposes the aromatic ring(s) orientation (diast. (2*S*,3*S*): $\chi_2 = +60^\circ$; diast. (2*S*,3*R*): $\chi_2 = -60^\circ$).

Ing was only synthesized in 1950 by Gagnon *et al.* as a racemic diastereoisomeric mixture², whereas Bfi diastereoisomers have never been prepared. The two α and β chiral centers first incited us to search for a strategy allowing α, β -stereocontrol. The method involving a β -diastereoselective 1,4-protonation of a 1-indanylenoyl sultam with L-Selectride^{3a}, followed by an α -stereocontrolled addition of a nitrosylchloride^{3b} proved unfruitful since hydride addition resulted in sultam cleavage^{3c}. In addition, attempting asymmetric alkylation of a sultam-derived glycine enolate synthon⁵ with chiral 1-indanylmethylate failed due to the partial epimerization of the electrophile by the chiral Schiff base⁴. These results led us to give up β -stereocontrol and to achieve the synthesis of these amino acids *via* a route which is depicted in the scheme.

The key step of this synthesis involved an asymmetric alkylation of the sultam-derived glycinate synthon **1**^{5b} with achiral 1-indanyl bromide^{6a} and the novel 1-benz[f]indanyl bromide. These electrophiles were obtained quantitatively by simple hydrobromination^{6b} of indene and benz[f]indene, respectively. The latter, whose preparation on a large scale was not reported until 1990^{7a-b}, was synthesized according to Burger's procedure^{7a}, except for the last step^{7c}. Since these halides show a high tendency to eliminate HBr, addition reactions with them require a carbanion possessing both nucleophilic and weakly basic properties⁸. However, they occurred readily with synthon **1**, probably due to a high delocalization of the carbanion charge. Thus, deprotonation of **1** with *n*-BuLi (THF, -78°C) followed by the addition of electrophile (HMPA, $-20^\circ\text{C} \rightarrow \text{r.t.}$) afforded, after workup and

flash-chromatography over basic alumina, Schiff base **2a-d** as a (2*S*,3*S*) and (2*S*,3*R*) mixture of diastereoisomers (scheme, table)⁹, whose structural features consist of a quasi-total (d.e.>95%) (*S*)- α -configuration and a slight β -stereocontrol, dependent on the electrophile¹⁰.

Scheme



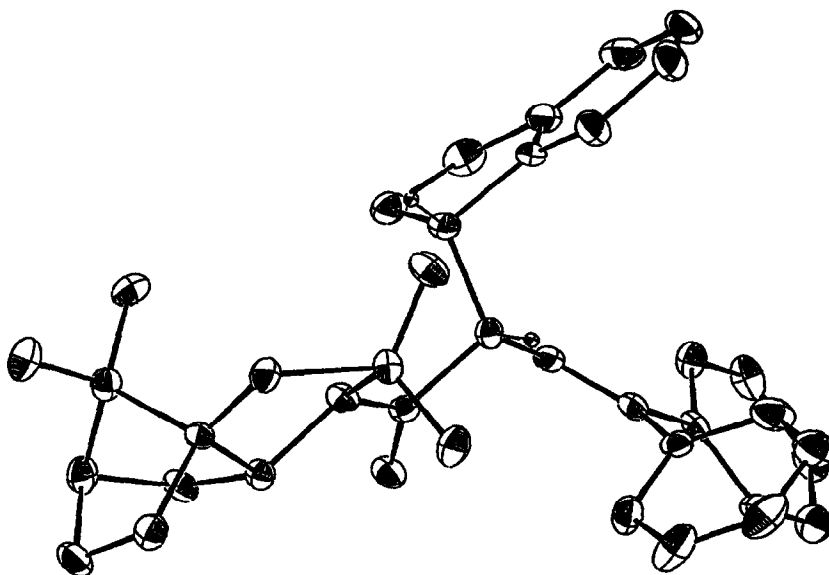
Hydrolysis of the imine function was carried out under ultrasonic irradiation (0.5N HCl, Et₂O, CHCl₃, 4h), to provide diastereoisomeric mixture of *N*-aminoacylsultam hydrochlorides. A smooth transformation into their free-amine counterparts with propylene oxide in diethylether allowed their separation by chromatography, to give in good yield enantiomerically pure *N*-aminoacylsultams (2*S*,3*S*) **3a,c** and (2*S*,3*R*) **3b,d** after recrystallization (CH₂Cl₂/Et₂O).

Table. Preparation of L-1-indanylglycines and L-1-Benz[*f*]indanylglycines **1** → **5**

R	Config.	Yield ^a) (%)			d.e. (%)		M.p. (°C)			[α] _D ²⁰	
		2	3	5	2 (α)	2 (β)	3	4b)	5	4d)	5e)
a	(2 <i>S</i> ,3 <i>S</i>)	69	83	57	> 95	2	178-180	185-188 ^{c)}	foam	+53.9	+28.6
b	(2 <i>S</i> ,3 <i>R</i>)	83	83	63	> 95		150-151	227-230 ^{c)}	109-111	-9.7	+10.8
c	(2 <i>S</i> ,3 <i>S</i>)	80	74	52	> 95		209-210	192-194 ^{c)}	102-106	+90.9	+112.7
d	(2 <i>S</i> ,3 <i>R</i>)	72	72	62	> 95	33	207-208	221-225 ^{c)}	213-216 ^{c)}	-69.9	-96.2

a) Pure recrystallized compound. b) HCl salts. c) Decomp. d) α_{obs} from TFA salts (c=5, AcOH) e) c=5, a-c: MeOH; d: DMF.

An X-ray analysis¹¹ of (2*S*,3*R*) Ing Schiff base **2b**, obtained from transimination of **3b** hydrochloride with benzophenone imine¹², allowed us to ascertain the β -configuration of the four diastereoisomer derivatives of Ing and Bfi **3a-d** (figure)¹³.

Figure. An ORTEP stereoview of (2*S*,3*R*) Ing precursor 2b

Clean removal of the sultam group proved to be highly dependent on the β -configuration: it could be accomplished readily with (2*S*,3*R*) 3b,d compounds according to the classical method, i.e. LiOH-THF/H₂O. But an important side-reaction, involving cleavage of the sultam ring, previously encountered with L-diphenylalanine^{5b}, was observed in the case of (2*S*,3*S*) 3a,c intermediates, even when phase-transfer conditions^{5b} were applied. Using sodium sulfide in THF/H₂O, this side-reaction was reduced to 10-15%. The crude amino acids hydrochlorides 4a-d were then obtained after extraction of sultam and acidification.

N-protection of these crude products under ultrasonic irradiation according to Luche's procedure¹⁴ (Boc₂O, NaHCO₃, EtOH) afforded, after flash chromatography and recrystallization, Boc-amino acids 5a-d: their high enantiomeric purity (e.c.>95%) was revealed on insertion in position 7 or 8 of the sequence of Substance P (Arg-Pro-Lys-Pro-Gln-Gln-Phe-Phe-Gly-Leu-Met-NH₂) by solid-phase synthesis followed by HPLC analysis of the resulting crude peptide. A TFA-deprotection could provide amino acids 4a-d.

In summary, we have shown that the sultam-derived glycine synthon can be alkylated by several base-sensitive electrophiles to achieve the synthesis of four aromatic α,β -chiral side-chain constrained α -amino acids. Use of the topographic properties of these probes to model the envelope of bioactive peptide is currently underway.

Acknowledgments: Financial support of this work by Rhône-Poulenc Rorer and Association Claude Bernard, Paris, is gratefully acknowledged. The authors thank Dr. Waissermann (Laboratoire de Chimie des Métaux de Transition, Université Pierre-et-Marie Curie, Paris) for carrying X-ray analysis.

REFERENCES AND NOTES

1. R. Dharanipragada, K.C. Russell, E. Nicolas, G. Toth, V.J. Hruby, "Proceedings of the Eleventh American Peptide Symposium", J.E. Rivier, G.R. Marshall, ESCOM, 1990, 937.
2. P.E. Gagnon, J.L. Boivin, P.A. Boivin, *Can. J. Research*, 1950, 28B, 207-212.
3. a) W. Oppolzer, G. Poli, *Tet. Lett.*, 1986, 27(39), 4717-4720; b) W Oppolzer, O. Tamura, *Tet. Lett.*, 1990, 31(7), 991-994; c) similar 1,2-additions were observed with Et₃SiH/TFA and MeMgCl. 1,4-additions were only observed with *Gilman* reagents R₂CuLi/PBu₃ (d.e. = 60%).
4. 1-indanylmethylate was even less reactive than 1-indanylbromide (several days were needed, at room temperature).
5. a) W. Oppolzer, R. Moretti, S. Thomi, *Tet. Lett.*, 1989, 30, 6009-6011; b) H. Josien, A. Martin, G. Chassaing, *Tet. Lett.*, 1991, 32(45), 6547-6550.
6. a) C. Courtot, A. Dondelinger, M.A. Haller, *Comptes Rendus de l'Acad. des Sciences*, 1924, 179, 1168-1171; b) We also prepared quantitatively 1-indanylbromide from 1-indanol with PPh₃/Br₂.
7. a) U. Burger, P.J. Thorel, J.P. Schaller, *Tet. Lett.*, 1990, 3155-3156 and references herein; b) L.A. Carpino, Y.Z. Lin, *J. Org. Chem.*, 1990, 55, 247-250; c) transformation of *endo+exo*-4,9-epoxy-3a,4,9,9a-tetrahydro-1H-benz[f]indene to benz[f]indene was achieved with 10% H₂SO₄ (reflux, 5h, 90%), i.e. same reagent used by Carpino *et al.* to deshydrate 1-benz[f]indanol.
8. These limitations might explain the low yield obtained by Gagnon *et al.*² in the racemic series of Ing using alkylcyanoacetic esters as carbanions. Similarly, we observed that N-acetyl sultam gave only elimination reaction with 1-indanyl bromide.
9. All new compounds were characterized by ¹H NMR; microanalysis gave satisfactory results.
Preparation of the electrophiles: pure indene, or benz[f]indene dissolved in methylene chloride, was saturated at 0°C with HBr gas. Excess HBr was removed from 1-benz[f]indanyl bromide by concentrating to dryness its methylene chloride solution three times, and from 1-indanyl bromide by four freezing-melting cycles with liquid nitrogen *in vacuo*. These electrophiles should be used quickly.
Alkylation (1 → 2): n-BuLi (1.6 N hexane, 1 equiv.) was added over 10 min to a solution of 1 in THF (4ml/mmol) and HMPA (1 ml/mmol) at -78°C under argon. Stirring the mixture to -20°C, injection at this temperature of the electrophile in THF (1 ml/mmol), then 2 h, r.t., 2 h, quenching with AcOH (THF), addition of Et₂O, washing with sat. aq. NH₄Cl, drying on magnesium sulfate, gave crude 2 which was flash-chromatographed over basic alumina (2a-b: cyclohexane/AcOEt 8:2 then CHCl₃; 2c-d: cyclohexane/AcOEt 7:3 then CHCl₃) and recrystallized (CH₂Cl₂/Et₂O).
Hydrolysis; separation of the diastereoisomers (2 → 3): a mixture of 2 in Et₂O (1.5 ml/mmol), CHCl₃ (1.5 ml/mmol), and 0.5N HCl (1.5 equiv.) was sonicated in a cleaning bath for 4 h. Concentration to dryness, trituration in Et₂O, filtration, afforded crude 3 hydrochlorides. Stirring in Et₂O with propylene oxide, 24 h, *in vacuo* removal of solvents, chromatography over silica gel in a long column (200g for 1.5g crude; h = 1.5m; Ø = 3cm; 3a-b: CH₂Cl₂/AcOEt 85:15; 3c-d: CH₂Cl₂/AcOEt 88:12) provided first (2S,3S) 3a,c, then (2S,3R) 3b,d, recrystallized in CH₂Cl₂/Et₂O.
Saponification; N-protection (3 → 4 → 5): Na₂S (1.5 equiv.) in H₂O (3 ml/mmol) was added at 0°C to 3 in THF (3 ml/mmol). After vigorous stirring at room temperature (3a: 24h; 3c: 8h; 3b,d: 5h), addition of water, cautious extraction of sultam with CHCl₃, acidification of the aqueous phase (pH=1-2), concentration to dryness, crude 4 was obtained. Addition of EtOH (8 ml/mmol), NaHCO₃ (0.40 g/mmol) and Boc₂O (1 equiv.), sonication (5a,b: 5h; 5c,d: 8h), filtration over celite, flash-chromatography over silica gel (5a-c: CH₂Cl₂/MeOH/AcOH 95:5:0.5; 5d: CHCl₃/MeOH/AcOH 90:10:0.5) and recrystallisation (5b: CH₂Cl₂/pentane; 5c: Et₂O/pentane; 3d: CH₂Cl₂/MeOH) afforded pure 5a-d.
10. β-diastereoisomers excess was inferred from the analysis of ¹H NMR spectra which exhibited important anisotropic effects of the aromatic rings on Me (sultam), H_α, H_β (and ³J_{αβ}) and H_γ, particularly.
11. Crystallographic data have been deposited at the *Cambridge Crystallographic Data Centre*.
12. M.J. O'Donnell, R.L. Polt, *J. Org. Chem.*, 1982, 47, 2663-2666.
13. β-configurations in Bfi serie were deduced from Ing's ones by comparison of ¹H NMR spectra between all (2S,3S) and (2S,3R) 2-5 Ing and Bfi derivatives, which showed strong analogies, or identity, of the coupling constants and the chemicals shifts.
14. J. Einhorn, C. Einhorn, J.L. Luche, *Synlett.*, 1991, 37.